

# Precision Gas Products Inc.

2455 Cawthra Road, Unit 21 Mississauga, Ontario, L5A 3P1

Tel: (905)-949-2626/1-888-730-8196

Fax: (905)-949-2688

Emergency Contact: Chemtrec (800) 424-9300

## Hydrogen Sulfide in Air

0.1 ppm to 2 %

## MATERIAL SAFETY DATA SHEET

### Identification

Revision Date 01-01-12

Products Name: HYDROGEN SULFIDE IN AIR 0.1PPM TO 2%

CAS Number: N/A

Chemical Family: Gas Mixture

Chemical formula: H<sub>2</sub>S in Air

MSDS identification Code/ Number: MSDS 105

### Composition/ Information on Ingredients

**Concentration**  
**Percent by Weight**  
**< .0001 to 2.0**

#### Ingredient Name

**HYDROGEN SULFIDE** CAS Number: 7783-06-4

#### Exposure Limits

- ACGIH TLV-TWA: 10 ppm
- ACGIH TLV-STEL: 15 ppm
- OSHA Final PEL-TWA: 10 ppm
- OSHA Final PEL-TWA: 15 ppm
- OSHA Trans. PEL-Ceiling: 20 ppm
- OSHA Trans. PEL-Peak: 50 ppm

**AIR** ACGIH TLV-TWA:D

98.0 to 99.995

CAS Number: 25635-88-5

### Hazard Identification

#### Hydrogen Sulfide:

At 0.3 to 30 ppm the odor is obvious and unpleasant.

At 50 ppm there may be eye irritation. Dryness and irritation of nose and throat.

**All Hydrogen Sulfide Gas Mixtures made at PGP are 50 ppm or less.**

### First Aid Measures

#### Eyes

Persons with potential exposure should not wear contact Lenses. Flush contaminated eyes with copious quantities of water. Part eyelids to assure complete flushing. Continue for 15minutes.

#### Skin

Flush affected areas with water. Remove contaminated clothing as rapidly as possible. Seek immediate medical attention.

#### Inhalation

Prompt medical attention is mandatory in all cases of overexposure. Rescue personnel should be equipped with self-contained breathing apparatus. Victims should be assisted to an uncontaminated area and inhale fresh air. Quick removal from the contaminated area is most important. If breathing has stopped, administer artificial resuscitation and supplemental oxygen. Further treatment should be symptomatic and supportive.

---

### Fire Fighting Measures

---

#### Flammable Properties

**Flash Point:** N/A- Gas

Lower Explosive Limit (%): 4

Upper Explosive Limit (%): 74.5

- Fire and Explosion Hazard: None
- Extinguishing Media: Water, carbon dioxide, dry chemical
- Fire Fighting Instructions: If possible, stop flow of gas; use water spray to cool surrounding containers.

---

### Accidental Release Measures

---

Evacuate all personnel from affected areas. Use appropriate protective equipment. If leak is in user's equipment, be certain to purge piping with an inert gas prior to attempting repairs. If leak is in container or container valve, contact CHEMTREC location for emergency assistance.

---

### Handling and Storage

---

- Handling and Storage Precautions

Protect cylinders from physical damage. Store in cool, dry, well – ventilated area of noncombustible construction away from heavily trafficked areas and emergency exits. Do not allow the temperature where cylinders are stored to exceed 125°F. Cylinders should be stored upright and firmly secured to prevent falling or being knocked over. Use a “first in, first out” inventory system to prevent full cylinders being stored for excessive periods of time.

Valve protection caps must remain in place unless container is secured with valve outlet piped to use point. Close valve after each use and when the container is empty. Do not drag, slide or roll cylinders. Use a suitable hand truck for cylinder movement. Use a pressure-reducing regulator when connecting container to piping or systems. Do not heat cylinder by any means to increase the discharge rate of product from the cylinder.

Never carry a compressed gas cylinder or a container of a gas in cryogenic liquid form in an enclosed space such as a car trunk, van or station wagon. A leak can result in a fire, asphyxiation or toxic exposure.

---

### Exposure Controls/Personal Protection

---

Engineering Controls: Use local exhaust to prevent accumulation above the exposure limits. A laboratory type hood is suitable for handling small or limited quantities.

Eye/Face Protection: Safety goggles or glasses.

Skin Protection: Protective gloves of any material

Other/General Protection: Safety shoes, safety shower, eyewash fountain

Respiratory Protection: A Type C respirator with full-face piece equipped with an escape bottle or self-contained breathing apparatus should be available for emergency use, or when concentrations exceed exposure limits.

---

### Physical & Chemical Properties

---

Appearance: A colorless gas.

Odor: A rotten egg odor.

#### Basic Physical Properties

Vapor Density (Air = 1): 1.0 @ 0C

Solubility (H2O): Slight

Boiling Point: -317.8F      -194.3C

Vapor Pressure: Above critical Temperature

---

**Stability & Reactivity**

---

Stability: Stable

Conditions to Avoid (Stability): Heat, spark and open flame and other ignition sources.

Incompatible Materials: All flammable materials

Hazardous Polymerization: Will not occur

---

**Toxicological Information**

---

Eye Effects: Low concentrations of hydrogen sulfide will generally cause irritation to the conjunctiva. Repeated exposure to low concentrations is reported to cause conjunctivitis, photophobia, corneal bullae, tearing, pain and blurred vision.

Skin Effects: May irritate the skin upon contact.

Acute Oral Effects: Ingestion is considered unlikely. However, hydrogen sulfide will cause irritation of mucous membranes, causing a burning feeling with excess salivation likely. Irritation of the gastrointestinal tract may also occur.

Acute Inhalation Effects: Continuous exposure to low (15 to 50 ppm) concentrations of hydrogen sulfide will generally cause irritation to mucous membranes, and may also cause headache, dizziness or nausea. Olfactory fatigue or paralysis of smell is also possible; thus detection of hydrogen sulfide by its odor is not considered adequate. Higher concentrations (200 to 300 ppm) may result in respiratory arrest leading to coma or unconsciousness. Exposures for more than 30 minutes at concentrations greater than 700 ppm have been fatal.

Hydrogen sulfide should be regarded as highly toxic. Toxicologically, it reacts with enzymes in the bloodstream and inhibits cellular respiration resulting in pulmonary paralysis, sudden collapse and death.

Miscellaneous Toxicological Information

Carcinogenicity – NTP: No                      IARC: No                      OSHA: No

Medical Conditions Aggravated by Exposure: Blood Disorders

---

**Ecological Information**

---

Other Environmental Information

This product does NOT contain any ingredients which are regulated on the U.S. EPA List of Toxic Chemicals (40CFR 372), and is therefore not subject to release under Section 313 of EPCRA/SARA Title III.

---

**Disposal Considerations**

---

Do not attempt to dispose of waste or unused quantities. Return in the shipping container properly labeled, with any valve outlet plugs or caps secure and valve protection cap in place to Precision Gas Products for proper disposal.

---

**Transport Information**

---

Proper Shipping Name: Compressed Gas, N.O.S., (Air, Hydrogen Sulfide)

Hazardous Class: 2.2

CT (DOT) Identification Number: UN 1956

CT (DOT) Shipping Label: Nonflammable gas

---

**Regulatory Information**

---

**SARA Title III Notifications and Information**

**SARA Title III – Hazard Class:** Acute Health Hazard

Chronic Health Hazard

Fire Hazard

Sudden Release of Pressure Hazard

---

**Other Information**

---

Hazard Rating	Health:	4 Extreme
	Fire:	4 Extreme
	Reactivity:	3 High

MSDS Identification Code/Number: MSDS 105

---

**Reference Documentation**

---

Compressed gas cylinders should not be refilled except by qualified producers of compressed gases. Shipments of a compressed gas cylinder, which has not been filled by the owner or with his (written) consent is a violation of Federal Law (49CFR).

---

**Disclaimer of Expressed & Implied Warranties**

---

Although responsible care has been taken in the preparation of the document, we extend no warranties and make no representations as to the accuracy or completeness of the information contained therein, and assume no responsibility regarding the suitability of this information for the user's intended purposes or for the consequences of this use. Each individual should make a determination as to the suitability of the information for their particular purpose(s).